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On the enzymic conversion of Δ^4 -3-hydroxy steroids to Δ^4 -3-ketones

Although the occurrence of Δ^4 -androstene- 3β , 17β -diol or Δ^4 -androstene- 3α , 17β -diol in natural sources has never been reported, these steroids are readily converted to testosterone by rat and chick liver homogenates. The enzyme activity was first described as being principally in the $5,000 \times g$ supernatant of rat liver homogenates and of rat and chick liver acetone powders. Further studies show that the activity is found in the soluble fraction following centrifugation at $78,000 \times g$. The activity is precipitated from the supernatant with ammonium sulfate between 0.4 and 0.55 saturation.

The pH optimum for the conversion of the Δ^4 -3-hydroxyl to the Δ^4 -3-ketone was between pH 7.5 and pH 8.3 using either a phosphate or tris(hydroxymethyl)aminomethane buffer. In the subsequent metabolism of the formed testosterone, greater amounts of Δ^4 -androstene-3,17-dione and reduced ring-A saturated 17-ketosteroids were-also formed at the higher pH ranges. Since the pH optimum for 17-ketosteroid formation was shown previously to be 8.3 by SWEAT et al.², the increased formation of these end products can be ascribed to the optimal pH for the formation of testosterone as well as for the 17-ketones.

Both DPN and TPN were equally effective in the conversion of the Δ^4 -3-hydroxyl to the Δ^4 -3-ketone for the more purified tissue preparations, although DPN was more effective for the 5,000 \times g supernatant preparation. An absolute requirement for the pyridine nucleotides has not been demonstrated, since tissue incubated without added pyridine nucleotide still contains some activity.

Incubation of the 3a- and 3β -isomers of Δ^4 -pregnenolone *8 with the 78,000 \times g supernatant fluid of rat liver resulted in the formation of progesterone. The incubation product was extracted with ethyl acetate and chromatographed on paper in the ligroin-propylene glycol system for 6 h. The major zone gave an orange color with dinitrophenyl hydrazine, a blue color with the Zimmermann reagent, and absorbed in the ultraviolet light at 240 m μ . The zone was chromatographed on silica gel, and crystals were eluted with benzene-ethyl acetate, 9:1. Recrystallization from aqueous methanol gave crystals, m.p. 124-8° C, which did not depress the melting point of authentic progesterone, m.p. 124-8° C. The identity of the substance as progesterone was verified by infrared analysis.

TABLE I the conversion of Δ^4 -androstene-3 β ,17 β -diol to testosterone by the supernatant of rat liver acetone powder

Incubation volume of 5 ml consisted of 0.002 M Δ^4 -androstene-3 β ,17 β -diol, 0.003 M pyridine nucleotide, and 2 g equivalent tissue wet weight in 0.1 M phosphate buffer, pH 7.4. The flasks were incubated in air at 37° for 30 min.

Tissue preparation	Pyridine nucleotide added	Percent conversion of
5,000 \times g super	+ DPN + TPN	42.3 20.9
		8.6
78,000 \times g super	+ DPN	45.6
	+ TPN	47.8

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³ M. Gut, J. Org. Chem., 21 (1956) 1327.

^{*} The 3a- and 3 β - Δ 4-androstenediols and the 3a- and-3 β - Δ 4-pregnenolones were generously supplied by Dr. Marcel Gut.